# AMENDMENTS ON SALINITY AND WATER RETENTION OF SAND BASE ROOTZONE AND TURFGRASS YIELD

# Rahayu<sup>\*1</sup>, Yang Geun Mo<sup>2</sup>, and Choi Joon Soo<sup>2</sup>

<sup>1</sup>Department of Soil Science, Faculty of Agriculture, Sebelas Maret University, Surakarta <sup>2</sup>Department of Green Landscape Architecture Science, Bioresource Science, Dankook University, Korea Submitted : 2019-02-20 Accepted : 2019-06-27

### ABSTRACT

This research was column pot experiment with turfgrass was Kentucky bluegrass (*Poa pratensis*) plant irrigated saline irrigation and the column soaked in saline water. Rootzone profile consisted of 20 cm using saline lake dredged up sand. The sand amendments of the root zone were soil, zeolite, bottom ash, and peat. The mixtures of topsoil were; 90% sand + 10% peat moss, 80% sand + 10% soil + 10 % bottom ash, 80% sand + 20% soil, 90% sand + 5% peat + 5% zeolite, and 80% sand + 20% bottom ash. Interruption layer with coarse sand with diameters over 2 mm of 20 cm and 10 cm loamy soil as the bottom layer of the column. The result showed that Kentucky bluegrass could grow in sand based growing media amended by peat, sandy loam soils, bottom ash and zeolite being irrigated by 2 dS m<sup>-1</sup> saline water. Sand-based growing media amended by peat resulted in the highest clipping weigh but showed the highest salt accumulations. Sand amended by bottom ash and applied gypsum decreased clipping weigh, decreased SAR and increased calcium (Ca) when compared to the soil + peat (SP). Sand amended by zeolite and gypsum decreased clipping weight, decreased sodium adsorption ratio (SAR) and higher Ca. Higher soil moisture retention of growing media promoted the growth of Kentucky bluegrass in spring, and lower moisture content promoted the growth in summer and fall season.

Keywords: amendment, salinity, turfgrass, water content

**How to Cite**: Rahayu, Mo Y. G., and Soo C. J. (2019). Amendments on Salinity and Water Retention of Sand Base Rootzone and Turfgrass Yield. Sains Tanah Journal of Soil Science and Agroclimatology, 16(1): 103-111 (doi: 10.20961/stjssa.v16i1.28132)

## Permalink/DOI: http://dx.doi.org/10.20961/stjssa.v16i1.28132

#### INTRODUCTION

Rapid urban population growth pressed land and fresh-water usage, thus golf course development was often pushed into non-fertile land with limited potable water. Attention and study on saline soil and saline water irrigation became necessary for turfgrass establishment. Many reports stated that turfgrass irrigation is typically considered as a low priority on

\* Corresponding Author : Email: rahayu\_uns@yahoo.co.id freshwater use (Kjelgren, Rupp, & Kilgren, 2000; Marcum, 2006), and reused water is often the primary source of soluble salts (Silvertooth, 2001). Miyamoto, Chacon, Hossain, & Martinez (2005) reported that water with the salinity of 1.3–1.5 dS m<sup>-1</sup> can be used for irrigation. Also, saline water with ECw 0.7 dS m<sup>-1</sup> could be used for irrigation on saline soil with ECe range from 1.3 to 1.5 dS m<sup>-1</sup> (Mancino & Pepper, 1992). However, using saline water caused many problems, such as soil salinization, salt injury to the turf and decreased salt leaching potential (BarrettLennard, 2003). Saline irrigation can cause the loss of plant stand, limited water uptake by the plant and specific ion toxicity and nutritional unbalance (Corwin & Lesch, 2003). Irrigation using saline water was reported to cause the decline of turf quality (Qian, Wilhelm, & Marcum, 2001). Salinity is an abiotic stress that affects the plant's ability to grow, develop, and achieve its full genetic potential (Läuchli & Grattan, 2014). Salinity can reduce the turgor of epidermal cells in both mature and expanding tissue and in the short term, salinity reduces leaf elongation through osmotic effects on the turgor of expanding tissue of plant (Thiel, Lynch, & Läuchli, 1988). Salinity decreases leaf water content, stomatal conductance, leaf water potential, and turgor potential. Under the saline condition, grass with higher water content is more tolerant than lower water content. Salinity reduced osmotic and water potential thus plants escape from dehydration and finally total dry matter production of a plant (Ahmad, Azooz, & Prasad, 2013). Incorrect management in saline condition will result in salt accumulation in topsoil, causing the growth of the turfgrass to be unsustainable and the turf quality to be unacceptable. Commonly salt accumulation is affected by volume of water, water movement in infiltration, percolation and drainage, and evapotranspiration pattern (Silvertooth, 2001).

Zeolite has become a common soil ameliorate in sand based growing media. Zeolite is the hydrated aluminosilicate materials with high cation exchange capacity from 100-230 me 100g<sup>-1</sup> (Ok, Anderson, & Ervin, 2003). Huang & Petrovic (1994) reported that zeolite increased the water and nutrient holding a capacity of sand-based media used in golf course greens and sports fields. Zeolite was also applied to sand based growing media in saline condition. Soil also can be used as an amendment to sand base rootzone when the sand content should be ranged from 78 to 87% by weight for acceptable saturated hydraulic conductivity (McCoy, 2006). The other abundant and cheap amendment materials are bottom ash. Bottom ash (BA) is a waste of coal in power plant that represents 13 – 20% of the total ash remaining in the bottom of a coalfired boiler after combustion. The major constituent of BA are Ca, Al, Fe, Mg, K, Si, Na and Ti, where Ca, Fe, Mg, K, and Si are essential to plant nutrients (Korcak, 1995). Gypsum (CaSO<sub>4</sub>) has a role in improving flocculation, enhancing aggregate stability, increasing infiltration rate (Shainberg et al., 1989) and increasing Ca<sup>2+</sup> content and replacing Na<sup>+</sup>. For sand based growing media, peat is the most frequently used organic amendments in golf course construction, because they have benefits in reducing the soil bulk density, improving rootzone aeration, increasing soil moisture retention, and gradual release of water available to plant (Bigelow, Bowman, Cassel, & Rufty, 2001; Waltz, Quisenberry, & McCarty, 2003). The growth of Kentucky bluegrass showed high clipping dry weight in June and low in August, while the weight of rhizome, root, and thatch increased with growth progressed in Korea (Yoon & Lee, 1992). Increase of salinity caused root/shoot weight ratio to increase, where shoot growth decreased linearly in all levels of salinity and root growth was increased to a maximum point and then declined (Harivandi, Butler, & Wu, 1992).

#### MATERIALS AND METHODS

This research was a column pot experiment conducted in Cheonan, Korea for 16 months, from June to October 2010. Turfgrass was Kentucky bluegrass (*Poa pratensis*) irrigated by fresh water during first one month after planting and then was followed by saline irrigation. Salinity levels of irrigation water were ECw of 2.0 dS m<sup>-1</sup> Turfgrass be fertilized by complete fertilizer (11-5-7) and was applied 3 times per year with rate 4 g N m<sup>-2</sup> per each application. The column pot contained sandbased rootzone. Bottom of columns was soaked in pond saline water with 5-10 cm depth with ECw around 3-6 dS m<sup>-1</sup>. Rootzone profile consisted of 20 cm using sand from the bottom of the saline lake by dredged up way. The sand amendments of the rootzone were soil, zeolite, bottom ash, and peat. The amendments then are compared by application gypsum and no gypsum application. The mixtures of topsoil were; 90% sand + 10% peat moss, 80% sand + 10% soil + 10 % bottom ash, 80% sand + 20% soil, 90% sand + 5% peat + 5% zeolite, and 80% sand + 20% bottom ash. Interruption layer with coarse sand with diameters over 2 mm of 20 cm and 10 cm loamy soil as the bottom layer of the column. The bottom layer of the column was saline soil with pH 6.7 and ECe 5.1 dS m<sup>-1</sup>, with sand, silt and clay contents of 63.8%, 32.0%, and 4.9%, respectively. The bottom of the column was plastic net thus the column was holly thus the pond water entry capillary to the rootzone.

The clipping tissue water content, soil moisture content, Magnesium and calcium, and SAR was investigated. Moisture contents were measured every 1 day, 2 days, and 3 days after irrigation with Time Domain Reflectometry (TDR). Calcium and magnesium analyze were by using 2 steps. The first step was to remove the Ca and Mg from the soil complex. Ten grams of dry soil was mixed with 30 ml of ammonium acetate (1M pH 7) in a flask and were shaken at 180 rpm for 30 minutes, then were filtered and leached by 70 ml of ammonium acetate to get extract solution in the ratio 10:1 of ammonium acetate and soil. Total hardness (Ca + Mg) was analyzed by 10 ml of extract solution buffered by 5 ml ammoniac buffer (pH 10). The color indicator was eriochrome black T solution and titration of the solution was by 0.01 M EDTA. Calcium was analyzed by using 2 ml solution extract buffered by 2 ml NaOH (2 M), colored by calcon indicator and then titrated by EDTA 0.01 M solution. Subtracting the Ca in ppm from total hardness was the Mg in ppm.

#### RESULTS

In Table 1 the moisture retention was increased significantly when soil and bottom ash were mixed together to sand that showed in SoBa and SSoBaGp. Soil material when applied together with gypsum in sand growing media increased the media moisture content, that showed at SSo compared to SSoGp in the first year of the experiment, moisture content of SPZ at 3 days after irrigation was higher than 2 days after irrigation, suggesting that there was a capillary rise of water movement. Table 1 showed also that soil moisture contents of SP and SSo were higher than other treatments when not added. gypsum was

Table 1. The soil moisture content of sand base growing media with various amend	ment materials in
saline condition	

	The first year (day)							The second year (day)							
Top Soil	Summer			Fall				Spring			Summer			Fall	
TOP SOIL	1	2	2 3		2 3		1	2	31		2 3		1	2	3
								(%)							
SP <sup>z</sup>	8.6bcd	6.4 a-d	6.3 cd	8.5 b	7.1 bc	7.4 bc	14.8a	11.9 a	9.3a	11.32a	10.83a	7.77 a	10.13 a	8.90 ab	8.77ab <sup>x</sup>
SoBa	13.8ab	9.8 a	11.0 a	18.1 a	13.9 a	11.4 a	9.7ab	9.2ab	7.5ab	8.34ab	5.93 a	6.77 a	8.33 ab	7.85abc	8.33abc
SSo	6.4d	4.6 d	4.7 d	6.8 b	5.8 c	5.9 c	8.9 ab	8.8ab	6.6ab	8.07ab	5.47 a	7.97 a	10.58 a	10.43 a	10.00 a
SPZ	12.8abc	9.0abc	10.3ab	11.0 b	9.2abc	10.1ab	9.1ab	8.5ab	7.2ab	7.37ab	5.00 a	6.60 a	7.40 ab	6.90 bc	7.13abc
SBa	8.4bcd	6.4 a-d	6.5bcd	6.9 b	6.7 bc	6.7 bc	8.7 b	8.1 ab	6.6ab	6.34ab	4.77 a	6.30 a	7.13 ab	6.47 bc	6.78 bc
SPGp	7.7 bcd	5.8 bcd	5.6 d	7.4 b	6.8 bc	7.0 bc	8.8 b	8.2 ab	6.5ab	7.20b	5.23 a	6.77 a	9.05 ab	8.65abc	8.22abc
SSoBaGp	15.3 a	9.5 ab	11.9 a	13.5ab	11.6ab	11.5a	10.2ab	10.0ab	8.1ab	8.18ab	6.57 a	8.30 a	8.30 ab	8.07abc	8.23abc
SSoGp	12.7a-d	9.5 ab	9.9abc	9.4b	8.8abc	9.0bc	11.4ab	10.3ab	.4ab	9.70ab	6.77 a	9.13 a	9.37 ab	8.65 ab	8.90 ab
SPZGp	6.9 cd	5.3 cd	5.1 d	6.8b	6.1c	6.0c	7.6b	7.3b	5.9 b	6.52b	4.10 a	5.80 a	6.22 b	5.83 c	5.63 c
SBaGp	8.9 bcd	6.0abcd	6.1 cd	10.1b	8.0bc	7.5bc	9.5ab	9.0ab	7.2ab	7.27ab	4.97 a	6.47 a	7.82 ab	7.38 bc	6.93 bc

<sup>x</sup> Means within a column followed by the same letter are not significantly different based on LSD.

<sup>y</sup>dai= day after irrigation.

<sup>z</sup>S= sand; P=peat; Ba= bottom ash; So= soil; Z= zeolite; Gp=gypsum

With gypsum application, SSoBaGp and SSoGp have higher moisture content than the others. Moisture contents of 3 days after irrigation were 11.0, 10.4 and 11.9% in the first summer (first year), but they were decreased to 6.77, 6.60 and 8.30% in the second summer (second year).

Table 2 showed that the SSoGp growing media showed higher pH at the end of summer, fall, and also spring, where their pH values were close to 7.0. SP and SPGp showed a lower pH than the other growing media. Application of gypsum generally increased the pH of most growing media. Increased salinity from the fall season to spring was paralleled with the decrease of pH. The SSoGp growing media showed higher pH at the end of summer, fall and also spring. Application of gypsum increased the pH of all treatments in the summer of the first year and the spring of the second year except the peat. Application of gypsum in peat amendment decreased the pH of media.

Table 3 showed that in the first year, the Ca content of the soil was increased in growing media when gypsum was applied. In the falls of the first year and second year, SPZ and SPZGp showed Ca contents higher than the other treatments. In the spring of the second year, SSo and SSoGp showed higher Ca content than

**Table 2.** The pH of sand base growing mediawith various amendment materials insaline condition.

Toncoil	First year		Second y	ear	
Topsoil -	Summer	Fall	Spring	Summer	Fall
SPx	6.27 f	6.76 cd	6.68 b	6.0 f	6.2 f*
SSoBa	6.24 f	6.78 cd	6.50 e	6.5 e	6.2 f
SSo	6.42 ef	7.08 a	6.48 e	6.9 a	6.4 e
SPZ	6.56 de	6.95 b	6.53d e	6.7 cd	6.5 d
SBa	6.63 cde	6.74 d	6.54d e	6.6 d	6.7 bc
SPGp	6.51 de	6.62 e	6.58 cd	6.1 f	6.7 bc
SSo BaGp	6.70 bcd	6.85 c	6.65 bc	6.6 d	7.0 a
SSoGp	7.03 a	7.05 a	6.78 a	6.8 ab	6.8 b
SPZGp	6.86 ab	7.00 ab	6.83 a	6.6 d	6.6 c
SBaGp	6.79 bc	7.07 a	6.84 a	6.8 bc	6.6 c
Means within	n a column fo	llowed by 1	the same le	tter are not s	significantl

different based on LSD

x S= sand; P=peat; Ba= bottom ash; So= soil; Z= zeolite; Gp=gypsum.

the other treatments. After one season from the establishment, soil and zeolite showed a higher Ca content than the other treatments. During the excessive rain period of the summer of the second year, the soil and zeolite showed a higher Ca level than the other treatments. Fine material in bottom ash also cannot hold Ca in saline water irrigation condition, thus Ca contents in SBa and SBaGp were lower than any other treatments at the fall of the second year. Ca content in SSoGp was lower than SSo, suggesting that gypsum in soil may have increased leaching of Ca.

Table 4 showed that irrigation water with a high concentration of Na and Mg may have caused the accumulation of Mg.

Table	3.	The	calci	Jm	conter	nt	of	sar	nd	base
	g	rowir	ng	me	edia	w	ith		va	rious
	а	meno	dment	t	materi	als	5	in	5	aline
	C	ondit	ion.							

Year fi	rst year	Year second year													
Summer	Summer Fall Spring Su		Summer	Fall											
		(ppm)													
80.8 d	106.9 c	122.2 cd	83.5 d	169.7 aby											
102.9 cd	113.6 bc	141.6 bcd	110.2 bc	161.0 ab											
108.2 cd	114.2 bc	157.0 ab	140.3 a	173.7 ab											
105.5 cd	142.3 ab	130.3 bcd	117.6 abc	187.0 a											
82.2 d	110.2 c	110.2 d	92.9 ab	151.0 b											
165.7 ab	110.9 c	110.9 d	98.2 cd	159.0 ab											
175.7 ab	122.9 bc	141.6 bcd	126.9 ab	167.0 ab											
192.4 a	123.6 bc	185.0 a	126.9 ab	167.0 ab											
181.0 ab	165.7 a	148.3 bc	134.9 ab	184.4 a											
140.9 bc	125.6 bc	127.6 bcd	127.6 ab	149.6 b											
	Summer 80.8 d 102.9 cd 105.5 cd 82.2 d 165.7 ab 175.7 ab 192.4 a 181.0 ab 140.9 bc	Summer         Fall           80.8 d         106.9 c           102.9 cd         113.6 bc           108.2 cd         114.2 bc           105.5 cd         142.3 ab           82.2 d         110.2 c           165.7 ab         110.9 c           175.7 ab         122.9 bc           192.4 a         123.6 bc           181.0 ab         165.7 a           140.9 bc         125.6 bc	Summer         Fall         Spring           80.8 d         106.9 c         122.2 cd           102.9 cd         113.6 bc         141.6 bcd           108.2 cd         114.2 bc         157.0 ab           105.5 cd         142.3 ab         130.3 bcd           82.2 d         110.2 c         110.9 d           175.7 ab         122.9 bc         141.6 bcd           192.4 a         123.6 bc         185.0 a           181.0 ab         165.7 a         148.3 bc           140.9 bc         125.6 bc         127.6 bcd	Summer         Fall         Spring         Summer           80.8 d         106.9 c         122.2 cd         83.5 d           102.9 cd         113.6 bc         141.6 bcd         110.2 bc           108.2 cd         114.2 bc         157.0 ab         140.3 a           105.5 cd         142.3 ab         130.3 bcd         117.6 abc           82.2 d         110.2 c         110.2 d         92.9 ab           165.7 ab         122.9 bc         141.6 bcd         126.9 ab           175.7 ab         122.9 bc         141.6 bcd         126.9 ab           192.4 a         123.6 bc         185.0 a         126.9 ab           181.0 ab         165.7 a         148.3 bc         134.9 ab											

y Means within a column followed by the same letter are not significantly different based on LSD.

z S= sand; P=peat; Ba= bottom ash; So= soil; Z= zeolite; Gp=gypsum.

# Table 4. The magnesium content of sand basegrowingmediawithvariousamendmentmaterialsinsaline

(	condition.											
Topsoil	Year first	year	ear Year second year									
	Summer	Fall	Spring	Summer	Fall							
SPz	143.4 a	26.3 a	147.7 a	64.8 de	198.9 bcy							
SSoBa	174.6 a	10.1 ab	164.8 a	89.1 abcd	234.6 ab							
SSo	112.6 a	17.8 ab	135.7 a	97.2 ab	247.2 ab							
SPZ	175.0 a	10.9 ab	150.4 a	78.6 bcde	202.6 abc							
SBa	185.2 a	4.1 b	159.5 a	59.2 e	194.1 bc							
SPGp	140.6 a	7.7 b	165.2 a	68.1 cde	245.5 ab							
SSo BaGp	144.6 a	14.6 ab	155.7 a	103.3 a	237.0 ab							
SSoGp	136.5 a	14.2 ab	138.5 a	91.2 abc	156.0 c							
SPZGp	173.8 a	8.9 b	148.6 a	78.2 bcde	250.8 ab							
SBaGp	159.6 a	15.0 ab	133.8 a	72.5 cde	269.8 a							
V Maana with	in a caluman	fallowed by	+ + + + + + + + + + + + + + + + + + + +	lattor are not	cignificantly							

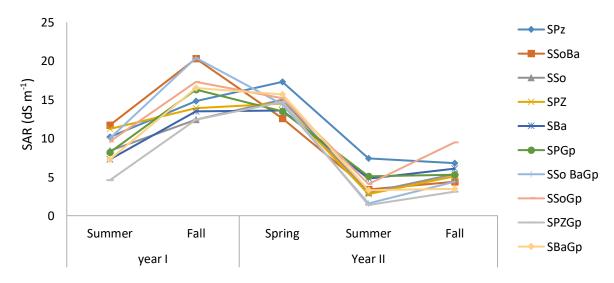
y Means within a column followed by the same letter are not significantly different based on LSD

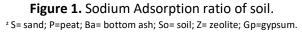
z S= sand; P=peat; Ba= bottom ash; So= soil; Z= zeolite; Gp=gypsum.

Decrease of Mg more than Ca in the fall of the first year and the summer of the second year may due to the different characteristic of both cation. At the summer of the first year, the application of gypsum did not increase the content of Mg in growing media. Summer of the second year was the period when most leaching occurred by frequent rain. In peat + zeolite modified soil, the Mg content was relatively consistent in the soil with little effect of gypsum addition. Figure 1 showed that SAR was increased from the summer to the fall of the first year, and decreased from the spring to the summer of the second year. In the first year of the experiment, SAR was generally higher in the fall season of dry weather. However, SAR was significantly decreased by high rainfall during summer. Addition of gypsum generally lowered SAR in most treatments.

Table 5 showed that total clipping fresh weight of SSoGp was higher than SP and SPGp, whereas the other treatments showed lower clipping yield than SP. In dry condition, the SP, SSo, SPGp, and SSoGp resulted in higher clipping fresh weight than the other treatments, while in wet condition SP, SPZ and SPGp resulted in higher clipping fresh weight. The SP, SSo, SPGp, and SSoGp resulted in higher clipping fresh weight than other treatments during fall and spring season. In the dry period, soil and peat addition resulted in higher clipping yield. Application of gypsum increased the clipping fresh weight of SSo and decreased the clipping dry weight of SPZGp. Application of gypsum increased the average clipping fresh weight of treatments such as SSo, SBa, and SsoBa.

Table 6 showed that the average of the tissue water content of Kentucky bluegrass was higher in SP than the other treatments. The tissue water content of Kentucky bluegrass fluctuated in all of the treatment with range 160 to 250 %. At fall season, SPZGp and SPZ have significantly lower tissue water content than the control, while SSoBa, SSo, and SSoGp showed higher tissue water content. This result suggests that soil can improve tissue water content, while zeolite declined tissue water content of Kentucky bluegrass. However, at spring season SPGp showed higher tissue water content, while SSoBa and SPZ showed lower tissue water content. Application of gypsum was no effect in tissue water content of Kentucky bluegrass. The maximum tissue water content was shown at the end of spring compare the other time of observation.





Topsoil	Sept O		Oct		Oct		Oct Nop I		May		June		July			
	1	15	1	15	1	4	19	1	15	30	20	4	20			
SP	57.4 a	24.3 a	24.3 ab	24.5 a	4.7 a	19.3 ab	38.0 a	55.8 a	27.1 abc	18.3 b	60.5 ab	68.6 a	45.95a			
SSoBa	56.5 a	22.6 a	17.2 bc	21.5 a	4.9 a	13.9 b	30.9 a	43.7 abc	24.8 abc	17.7 b	61.4 ab	53.8 bc	37.79a			
SSo	41.8 ab	21.7 a	23.5 abc	25.4 a	4.8 a	25.3 a	35.3 a	45.9 abc	25.8 abc	21.3 ab	60.6 ab	62.5 ab	40.91a			
SPZ	54.8 a	25.2 a	17.7 bc	22.6 a	3.1 a	14.4 b	29.7 a	35.9 c	23.4 c	21.5 ab	62.8 ab	57.4 bc	43.64a			
SBa	51.1 ab	21.7 a	23.5abc	22.3 a	3.4 a	14.6 b	30.0 a	37.3 bc	23.2 c	17.7 b	56.4 ab	46.5 c	40.72a			
SPGp	58.3 a	27.3 a	27.4 a	24.9 a	3.4 a	24.6 a	38.9 a	52.9 ab	30.3 a	17.2 b	61.6 ab	62.2 ab	39.07a			
SSoBaGp	50.0 ab	25.5 a	18.5 bc	18.8 a	3.8 a	15.2 b	36.2 a	44.7 abc	23.6 bc	18.4 b	53.5 b	51.3 bc	38.55a			
SSoGp	57.8 a	26.0 a	20.4abc	27.9 a	4.9 a	17.2 b	33.4 a	53.3 a	29.1 ab	28.6 a	66.2 a	60.6ab	35.90a			
SPZGp	35.9 b	19.6 a	15.8 c	20.2 a	3.6 a	18.7 ab	32.4 a	45.3abc	25.1abc	15.8 b	60.3 ab	53.5 bc	35.20a			
SBaGp	51.5 ab*	23.2 a	20.0 abc	24.4 a	5.0 a	18.7 ab	37.6 a	51.5 abc	22.7 с	19.7 ab	59.0 ab	55.0 bc	40.10a			

**Table 5.** Clipping fresh weight of Kentucky bluegrass at various soil amendments growing media in saline condition.

<sup>y</sup> Means within a column followed by the same letter are not significantly different based on LSD.

<sup>z</sup> S= sand; P=peat; Ba= bottom ash; So= soil; Z= zeolite; Gp=gypsum.

 Table 6. The leaf water content of Kentucky bluegrass at various soil amendments growing media in saline condition.

Toncoil	Sept	Sept (		Oct			May Jun		June Ju		July		Aug	
Topsoil -	1	15	1	15	30	4	19	1	15	30	20	4	19	
SP	250.6ab	207.3 a	201.4 a	211.6 a	184.5ab	201.4a	236.8a	217.9b	193.5 a	203.6ab	201.0a	210.1 a	287.2a	
SSoBa	279.8 a	211.0 a	198.0 a	207.8 a	176.9ab	170.3a	227.8a	219.7b	178.3 d	199.9ab	197.7ab	201.4ab	242.6a	
SSo	257.6ab	192.6 a	198.5 a	212.3 a	177.6ab	199.6a	232.8a	219.4b	181.5bcd	202.4 ab	196.3ab	201.1ab	250.9a	
SPZ	232.9ab	211.2 a	201.3 a	206.4 a	161.8 b	187.3a	233.0a	220.0b	190.5abc	205.8 ab	199.1ab	203.4ab	254.9a	
SBa	250.4 ab	191.7 a	204.6 a	205.0 a	172.1 ab	164.4a	223.1a	219.6b	179.7 cd	198.5 ab	195.4ab	189.0 b	213.8a	
SPGp	252.8ab	205.6 a	201.2 a	208.9 a	170.3ab	191.7a	236.0a	234.2a	191.8ab	234.9 a	201.0 a	202.7ab	201.4a	
SSoBaGp	254.5 ab	213.5 a	188.4 a	201.8 a	165.0 b	190.2a	228.7a	217.0b	182.0a-d	197.8 ab	200.4 a	196.6ab	228.9a	
SSoGp	259.2ab	213.1 a	192.7 a	209.9 a	168.9ab	163.8a	227.4a	216.1b	193.5 a	211.9 ab	205.8 a	203.2ab	237.9a	
SPZGp	214.1 b	189.1 a	190.6 a	199.1 a	171.9ab	201.1a	228.9a	218.7b	185.9bcd	170.7 b	197.9ab	196.0ab	205.6a	
SBaGpx	244.0ab	188.4 a	189.6 a	203.7 a	192.0 a	190.8a	240.8a	219.8b	181.0bcd	204.6 ab	184.9 b	192.2ab	225.0a	

<sup>y</sup> Means within a column followed by the same letter are not significantly different based on LSD.

<sup>z</sup> S= sand; P=peat; Ba= bottom

#### DISCUSSION

Refer to Table 1 the amendment of soil mixed with bottom ash and soil mixed to gypsum increased sand based medium moisture retention. This result correlates with Buck, CPSSc, & Labuz (2005) that soil and bottom ash has high silt content that can increase soil moisture storage. Also, increased salt accumulation can cause an increase in water content (Thompson, Gallardo, Fernández, Valdez, & Martínez-Gaitán, 2007). Table 1 also showed that zeolite amendment increased the soil moisture content of growing media, however soil moisture content was decreased when zeolite was applied together with gypsum. Al-Busaidi et al. (2008) reported that application of zeolite enhanced water content and water residence under saline treatment and salt holding capacity of the soil. The ability of zeolite in enhancing capillary rise movement was reported by Huang & Petrovic (1994); Wasura & Petrovic (2001), where it was deducted that zeolite can improve capillary water movement. In Table 1 SSoBaGp and SSoGp have higher moisture content than the other treatment. Buck et al. (2005) reported that bottom ash improves the permeability of soils. Improved permeability of soil can reduce the water content of the topsoil. Application of gypsum may increase the rate of infiltration and may increase field-saturated hydraulic conductivities (Ilyas, Miller, & Qureshi, 1993). Haisheng et al. (2008) reported that gypsum can increase Ca<sup>2+</sup> content and replace Na<sup>+</sup> and Mg<sup>2+</sup> to improve soil permeability.

Increased salinity from the fall season to spring was paralleled with the decrease of pH. Wilhelm, Alshammary, & Qian (2010) reported that soil pH was low with increased salinity by saline water irrigation. Application of gypsum in peat amendment decreased the pH of media. Applied gypsum increases conditions such as infiltration rate (Shainberg et al., 1989), soil permeability (Haisheng et al., 2008) and supply of soluble calcium (Mzezewa, Gotosa, & Nyamwanza, 2003). This enhanced condition by applied gypsum may have increased the decomposition of peat, resulting in the release of organic acid into the soil solution, thus decreasing the pH of the growing media (Table 1). Supply of Ca from gypsum addition may have also acted in buffering the pH of bottom ash from acidification by rainfall. Korean bottom ash used in this research was reported to contain high Ca level-up to 61% (Lim, Han, Ahn, & You, 2010). Johnson & Furrer (2002) also reported that bottom ash contains soluble salt such as calcite, which plays a predominant buffering constituent in time as soluble basic Ca salts.

Base on Table 3, during the excessive rain period, the soil and zeolite amendments showed a higher Ca level in the sand media. This result correlates with Al-Busaidi et al. (2008) that states the zeolite restrict nutrient and salt leaching. Bottom ash and peat resulted in lower Ca content from the summer of the first year to the summer of the second year. Even though the bottom ash amendment contains Ca, Mg and Na (Korcak, 1995), those elements are still in the primer mineral, thus bottom ash cannot work as an immediate source. Table 3 showed that Ca content in SSoGp was lower than SSo, suggesting that gypsum in soil may have increased leaching of Ca. Application of gypsum to soil increased the solubilities of Ca and Mg and decreased the solubility of Na. Since binding affinity is Na<sup>+</sup> <  $Mg^{2+}$  <  $Ca^{2+}$  with clay (Rytwo, Banin, & Nir, 1996), if the Ca was retained more by the clay, then the calcium content of the soil would have been higher. Haisheng et al. (2008) reported that gypsum as chemical modifying agents is the main component of saline-sodic soil amelioration, which can increase Ca<sup>2+</sup> content and replace Na<sup>+</sup> and Mg<sup>2+</sup> from soil colloids to improve soil condition and soil permeability. Supply of Ca from gypsum may be the main reason for lowering SAR. However, amendments of peat + zeolite + gypsum showed lower SAR than the other treatment in the fall season of the second year, was showed in Figure 1. At the summer of the second year when the salt was leached out from root zone, soil+ bottom ash + gypsum and peat +zeolite + gypsum showed lower SAR than the others, and significantly lower than the peat amendments. The high clipping fresh weight yield of SPZ and SPZGp in Table 5 showed that the ability of zeolite to improve turfgrass growth in a wet season. The effect of zeolite to improve the turfgrass growth may be related to CEC, nutrient retention and infiltration. Huang & Petrovic (1994) reported that zeolite increase nutrient holding capacity, CEC, and reduced nutrient leaching potential of sand based media (Wasura & Petrovic, 2001).

#### CONCLUSION

Soil, peat, and bottom ash showed beneficiary as amendment of sand based growing media in saline condition even though soil and bottom ash amendment showed lower clipping fresh wight. Peat as amendment resulted in high clipping weigh of turfgrass even though showed salt accumulations. Gypsum can increase Ca content and reduce SAR when added to bottom ash or zeolite amendment. Higher soil moisture retention of growing media promoted the growth of Kentucky bluegrass in spring, and lower moisture content promoted the growth in summer and fall season.

### REFERENCES

Ahmad, P., Azooz, M. M., & Prasad, M. N. V. (2013). Ecophysiology and Responses of Plants under Salt Stress (P. Ahmad, M. M. Azooz, & M. N. V. Prasad, Eds.). New York: Springer 10.1007/978-1-4614-4747-4

Al-Busaidi, A., Yamamoto, T., Inoue, M., Eneji, A. E., Mori, Y., & Irshad, M. (2008). Effects of Zeolite on Soil Nutrients and Growth of Barley Following Irrigation with Saline Water. *Journal of Plant Nutrition*, 31(7), 1159–1173.

10.1080/01904160802134434

- Barrett-Lennard, E. G. (2003). The interaction between waterlogging and salinity in higher plants: causes, consequences and implications. *Plant and Soil*, *253*(1), 35– 54. 10.1023/A:1024574622669
- Bigelow, C. A., Bowman, D. C., Cassel, D. K., & Rufty, T. W. (2001). Creeping Bentgrass Response to Inorganic Soil Amendments and Mechanically Induced Subsurface Drainage and Aeration. *Crop Science*, 41(3), 797–805.
- Buck, J. K., CPSSc, & Labuz, L. L. (2005). Bottom Ash Fines as a Soil Amendment for Turfgrass and Site Closure-Laboratory and Mesocosm Studies at PPL Brunner Island and Montour Steam Electric Station. Retrieved from http://www.flyash.info
- Corwin, D. L., & Lesch, S. M. (2003). Application of Soil Electrical Conductivity to Precision Agriculture. *Agronomy Journal*, *95*(3), 455–471. 10.2134/agronj2003.0455
- Haisheng, H., Wenjie, W., Hong, Z., Yuangang, Z., Zhonghua, Z., Yu, G., ... Xingyang, Y. (2008). Influences of addition of different krilium in saline-sodic soil on the seed germination and growth of cabbage. *Acta Ecologica Sinica*, 28(11), 5338–5346. 10.1016/S1872-2032(09)60009-3
- Harivandi, M. A., Butler, J. D., & Wu, L. (1992). Salinity and Turfgrass Culture. In *Turfgrass- Agronomy Monograph* USA: American Society of Agronomy, Crop Science Society of America, Soil Science Society of America (pp. 207–229). 10.2134/agronmonogr32.c6
- Huang, Z. T., & Petrovic, A. M. (1994). Physical Properties of Sand as Affected by Clinoptilolite Zeolite Particle Size and Quantity. *Journal of Turfgrass Management*, 1(1), 1–15. 10.1300/J099v01n01\_01
- Ilyas, M., Miller, R. W., & Qureshi, R. H. (1993). Hydraulic Conductivity of Saline-Sodic Soil

after Gypsum Application and Cropping.Soil Science Society of America Journal,57(6),10.2136/sssaj1993.03615995005700060

031x

- Johnson, C. A., & Furrer, G. (2002). Influence of Biodegradation Processes on The Duration of CaCO3 as a pH Buffer in Municipal Solid Waste Incinerator Bottom Ash. *Environmental Science* & *Technology*, 36(2), 215–220.
- Kjelgren, R., Rupp, L., & Kilgren, D. (2000). Water Conservation in Urban Landscapes. *Hort. Science*, *35*(6), 1037–1040.
- Korcak, R. F. (1995). Utilization of Coal Combustion by-Products in Agriculture and Horticulture. In D. L. Karlen, R. J. Wright, & W. O. Kemper (Eds.), Agricultural Utilization of Urban and Industrial By-Products. USA: American Society of Agronomy, Crop Science Society of America, Soil Science Society of America (pp. 107–130).
- Läuchli, A., & Grattan, S. R. (2014). Plant Abiotic Stress: Salt. In *Encyclopedia of Agriculture and Food Systems* (pp. 313–329). 10.1016/B978-0-444-52512-3.00171-6
- Lim, M., Han, G.-C., Ahn, J.-W., & You, K.-S. (2010). Environmental remediation and conversion of carbon dioxide (CO(2)) into useful green products by accelerated carbonation technology. *International Journal of Environmental Research and Public Health*, 7(1), 203–228. 10.3390/ijerph7010203
- Mancino, C. F., & Pepper, I. L. (1992). Irrigation of Turfgrass with Secondary Sewage Effluent: Soil Quality. *Agronomy Journal*, *84*(4), 650.
- Marcum, K. B. (2006). Use of saline and nonpotable water in the turfgrass industry: Constraints and developments. *Agricultural Water Management, 80*(1– 3), 132–146.
- McCoy. (2006). *Golf Course Soils and Water Science*. USA: Ohio State Univ.
- Miyamoto, S., Chacon, A., Hossain, M., & Martinez, I. (2005). Soil Salinity of Urban Turf Areas Irrigated with Saline Water I. Spatial Variability. *Landscape and Urban Planning*, 71(2–4), 233–241.

10.1016/j.landurbplan.2004.03.006

- Mzezewa, J., Gotosa, J., & Nyamwanza, B. (2003). Characterisation of A Sodic Soil Catena for Reclamation and Improvement Strategies. *Geoderma*, *113*(1–2), 161–175. 10.1016/S0016-7061(02)00337-3
- Ok, C.-H., Anderson, S. H., & Ervin, E. H. (2003). Amendments and Construction Systems for Improving the Performance of Sand-Based Putting Greens. *Agronomy Journal*, *95*(6), 1583–1590. 10.2134/agronj2003.1583
- Qian, Y. L., Wilhelm, S. J., & Marcum, K. B. (2001). Comparative Responses of Two Kentucky Bluegrass Cultivars to Salinity Stress. *Crop Science*, 41(6), 1895–1900. 10.2135/cropsci2001.1895
- Rytwo, G., Banin, A., & Nir, S. (1996). Exchange Reactions in the Ca-Mg-Na-Montmorillonite System. *Clays and Clay Minerals*, 44(2), 276–285. 10.1346/CCMN.1996.0440212
- Shainberg, I., Sumner, M. E., Miller, W. P., Farina, M. P. W., Pavan, M. A., & Fey, M.
  V. (1989). Use of Gypsum on Soils: A Review. In B. . Stewart (Ed.), Advances in Soil Science. New York: Springer (pp. 1– 111). 10.1007/978-1-4612-3532-3\_1
- Silvertooth, J. C. (2001). Saline and Sodic Soil Identification and Management for Cotton. Retrieved June 29, 2019, from University of Arizona website: https://cals.arizona.edu/crop/cotton/soil mgt/saline\_sodic\_soil.html

Thiel, G., Lynch, J., & Läuchli, A. (1988). Short-

term Effects of Salinity Stress on the Turgor and Elongation of Growing Barley Leaves. *Journal of Plant Physiology*, *132*(1), 38–44. 10.1016/S0176-1617(88)80180-9

- Thompson, R. B., Gallardo, M., Fernández, M.
  D., Valdez, L. C., & Martínez-Gaitán, C.
  (2007). Salinity Effects on Soil Moisture Measurement Made with a Capacitance Sensor. Soil Science Society of America Journal, 71, 1647–1657.
  10.2136/sssaj2006.0309
- Waltz, F. C., Quisenberry, V. L., & McCarty, L. B. (2003). Physical and Hydraulic Properties of Rootzone Mixes Amended with Inorganics for Golf Putting Greens. *Agronomy Journal*, *95*(2), 395–404.
- Wasura, J. P., & Petrovic, A. M. (2001). Physical stability of inorganic amendments used in turfgrass rootzones. *Int. Turfgrass Soc. J. Research*, *9*, 637–641.
- Wilhelm, S., Alshammary, S., & Qian, Y. (2010). Establishment, Growth and Irrigation Requirements of Kentucky Bluegrass and Tall Fescue as Influenced by Two Irrigation Water Sources. *Research Journal of Environmental Sciences*, 4(5), 443–451. 10.3923/rjes.2010.443.451
- Yoon, Y. B., & Lee, J. S. (1992). The Growth and Thatch Accumulation of Kentucky Bluegrasses As Affected By Cutting Management 1. Varietal Differences Under Removing Clipping Residues. *Korea Journal of Turfgrass Science*, 6(1), 29–37.